

LETTERS TO THE EDITOR

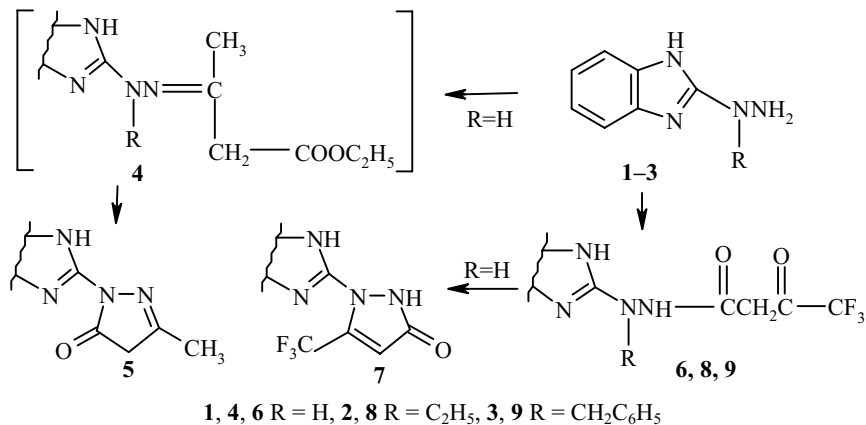
REACTION OF 2-HYDRAZINO- BENZIMIDAZOLES WITH ACETOACETIC ESTER AND TRIFLUOROMETHYL- ACETOACETIC ESTER

M. V. Povstyanoi, V. P. Kruglenko, and V. M. Povstyanoi

Keywords: 1-(2-benzimidazolyl)-1-ethyl-2-hydrazide of trifluoromethylacetoacetic acid, 2-hydrazinobenzimidazole, 2-(3-methyl-5-oxopyrazol-1-yl)benzimidazole, trifluoromethylacetoacetic ester, 2-(2H-3-oxo-5-trifluoromethylpyrazol-1-yl)benzimidazole.

Condensation of 2-hydrazino(alkylhydrazino)benzimidazoles **1-3** with β -dicarbonyl compounds occurs ambiguously. Depending on the nature of the dinucleophilic reagent either 2-substituted benzimidazoles [1] are obtained or ring closure occurs to give 1,2,4-triazepino- [2] and 1,2,4-triazinobenzimidazoles [3].

Condensation of 2-hydrazinobenzimidazole (**1**) with acetoacetic ester in boiling methanol (0.5 h) with a catalytic amount of hydrochloric acid occurs *via* initial formation of hydrazone **4** which cyclizes in the reaction conditions to 2-(3-methyl-5-oxopyrazol-1-yl)benzimidazole (**5**). Under analogous conditions hydrazine **1** reacts with trifluoromethylacetoacetic ester to give trifluoromethylacetoacetic acid 2-benzimidazolylhydrazide (**6**) and the product of its cyclization, isomeric 2-(2H-3-oxo-5-trifluoromethylpyrazol-1-yl)benzimidazole (**7**).



Kherson State Technical University, Kherson 73008, Ukraine; e-mail: kstu@cherson.ua. Translated from *Khimiya Geterotsiklicheskikh Soedinenii*, No. 1, pp. 127-128, January, 2001. Original article submitted November 3, 2000.

Condensation of alkylhydrazinobenzimidazoles **2** and **3** with trifluoromethylacetoacetic ester in boiling acetic acid (3 h) or toluene (5 h) stopped at the stage of synthesis of the substituted hydrazides **8** and **9**; attempts of cyclization of these compounds were unsuccessful.

2-(3-Methyl-5-oxopyrazol-1-yl)benzimidazole (5). Yield 60%; mp 249-250°C (acetonitrile). R_f 0.71. IR spectrum (KBr), ν , cm^{-1} : 1660 (C=O), 3330 (NH). ^1H NMR spectrum (DMSO- d_6), δ , ppm: 1.28 (3H, s, CH_3); 2.19 (2H, s, CH_2); 7.17-7.52 (4H, m, H_{arom}). Mass spectrum, m/z (I_{rel} , %): 215 (10), M^+ 214 (100), $[(\text{M} + \text{H}) - \text{C}_4\text{H}_5\text{N}_2\text{O}]^+$ 118 (28), $[\text{M} - \text{C}_7\text{H}_5\text{N}_2]^+$ 97 (18). Found, %: C 61.83; H 4.65; N 26.45. $\text{C}_{11}\text{H}_{10}\text{N}_4\text{O}$. Calculated, %: C 61.71; H 4.74; N 26.15.

Hydrazide 6. Yield 10%; mp 264-266°C (DMF). IR spectrum (KBr), ν , cm^{-1} : 1670, 1710 (C=O), 3250 (NH). Found, %: C 45.99; H 3.21; N 19.60. $\text{C}_{11}\text{H}_9\text{N}_4\text{F}_3\text{O}_2$. Calculated, %: C 46.15; H 3.17; N 19.57.

2-(2H-3-Oxo-5-trifluoromethylpyrazol-1-yl)benzimidazole (7). Yield 80%; mp 241-242°C (aqueous methanol). R_f 0.84. IR spectrum (KBr), ν , cm^{-1} : 1670 (C=O), 3260 (NH). ^1H NMR spectrum (DMSO- d_6), δ , ppm: 5.31 (1H, s, NH); 7.30-7.60 (4H, m, H_{arom}). Mass spectrum, m/z (I_{rel} , %): 269 (14), M^+ 268 (100), $[\text{M} - \text{F}]^+$ 249 (3), $[\text{M} - \text{CF}_3]^+$ 199 (27), $[(\text{M} + \text{H}) - \text{NHCOCHCCF}_3]^+$ 132 (12), $[(\text{M} + \text{H}) - \text{C}_4\text{H}_2\text{N}_2\text{F}_3\text{O}]^+$ 118 (24). Found, %: C 49.26; H 2.63; N 20.89. $\text{C}_{11}\text{H}_7\text{N}_4\text{F}_3\text{O}$. Calculated, %: C 49.26; H 2.63; N 20.89.

Trifluoromethylacetoacetic Acid 1-(2-Benzimidazolyl)-1-ethyl-2-hydrazide (8). Yield 80%; mp 230-231°C (aqueous methanol). IR spectrum (KBr), ν , cm^{-1} : 1660, 1720 (C=O), 3190 (NH). ^1H NMR spectrum (DMSO- d_6), δ , ppm: 1.20 (3H, t, CH_2CH_3); 3.37 (2H, s, CH_2); 3.78 (2H, q, CH_2N); 6.90-7.40 (4H, m, H_{arom}). Found, %: C 49.41; H 4.43; N 17.85. $\text{C}_{13}\text{H}_{13}\text{N}_4\text{F}_3\text{O}$. Calculated, %: C 49.69; H 4.13; N 17.83.

Trifluoromethylacetoacetic Acid 1-(2-Benzimidazolyl)-1-benzyl-2-hydrazide (9). Yield 88%; mp 220-221°C (acetonitrile). IR spectrum (KBr), ν , cm^{-1} : 1650, 1680 (C=O), 3180 (NH). ^1H NMR spectrum (CD_3OD), δ , ppm: 3.20 (2H, s, CH_2); 4.82 (2H, s, CH_2N); 6.88-7.25 (4H, m, H_{arom}). Found, %: C 57.32; H 3.95; N 14.95. $\text{C}_{18}\text{H}_{15}\text{N}_4\text{F}_3\text{O}$. Calculated, %: C 57.44; H 4.01; N 14.88.

REFERENCES

1. N. A. Klyuev, M. V. Povstyanoi, G. G. Aleksandrov, and V. P. Gumennii, *Khim. Geterotsikl. Soedin.*, 88 (1983).
2. M. V. Povstyanoi, V. P. Kruglenko, E. N. Fedosenko, and N. A. Klyuev, *Khim. Geterotsikl. Soedin.*, 1065 (1990).
3. V. P. Kruglenko, E. N. Fedosenko, and M. V. Povstyanoi, *Ukr. Khim. Zh.*, **56**, 1089 (1990).